



Dipartimento di  
Scienze Chimiche



Università degli  
Studi di Padova



Dipartimento di  
Fisica e Astronomia

CORSO DI LAUREA IN SCIENZA DEI MATERIALI

Tesi di laurea

# L'ANIDRIDE CARBONICA: DA GAS SERRA A RISORSA

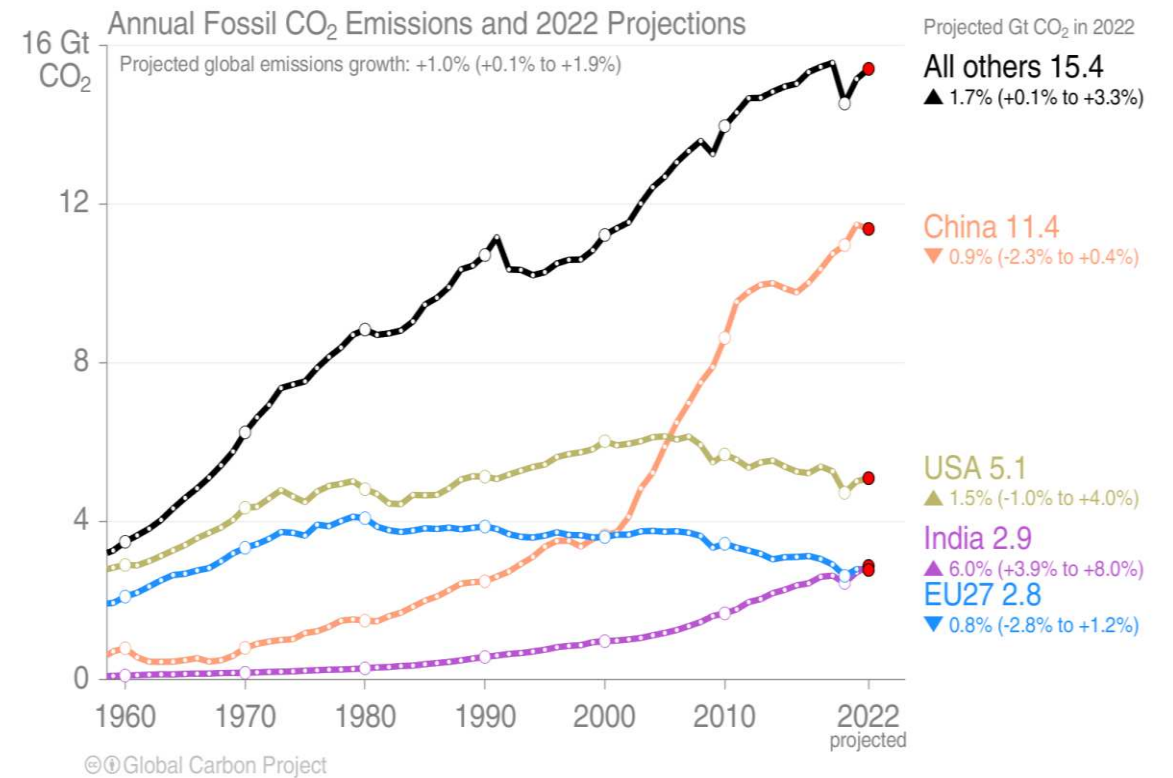
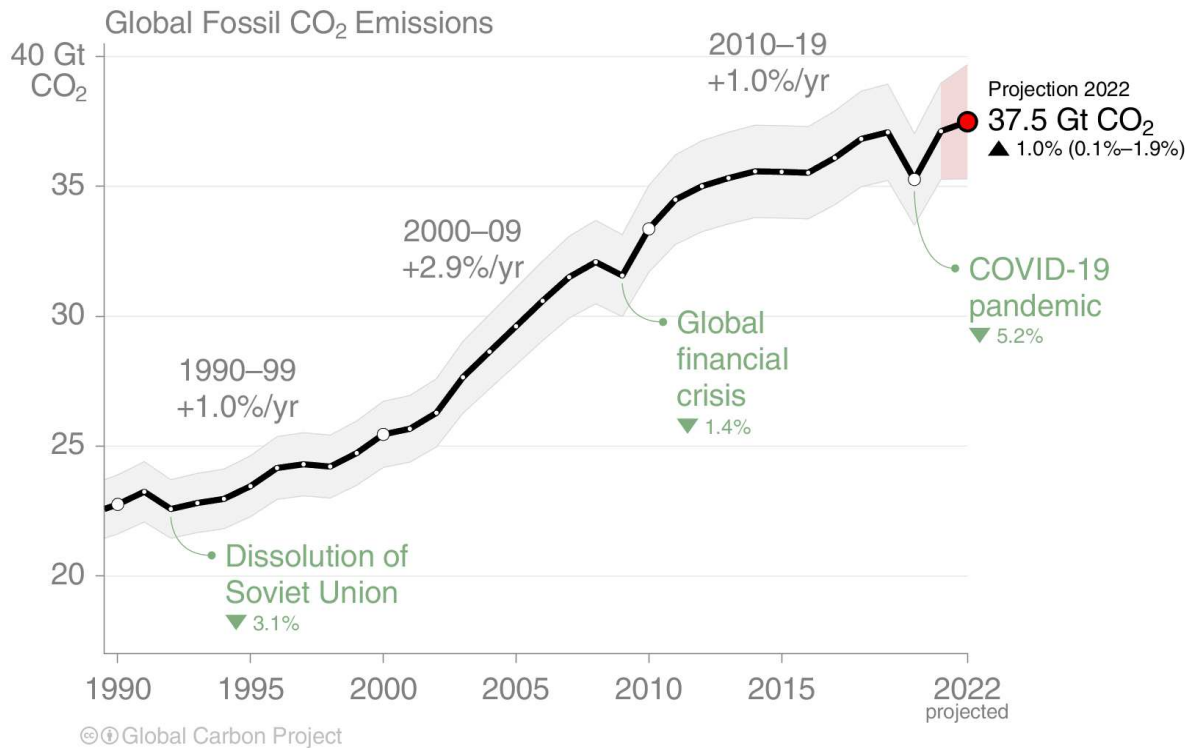
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Anno accademico: 2022/2023

# EMISSIONI GLOBALI DI CO<sub>2</sub>

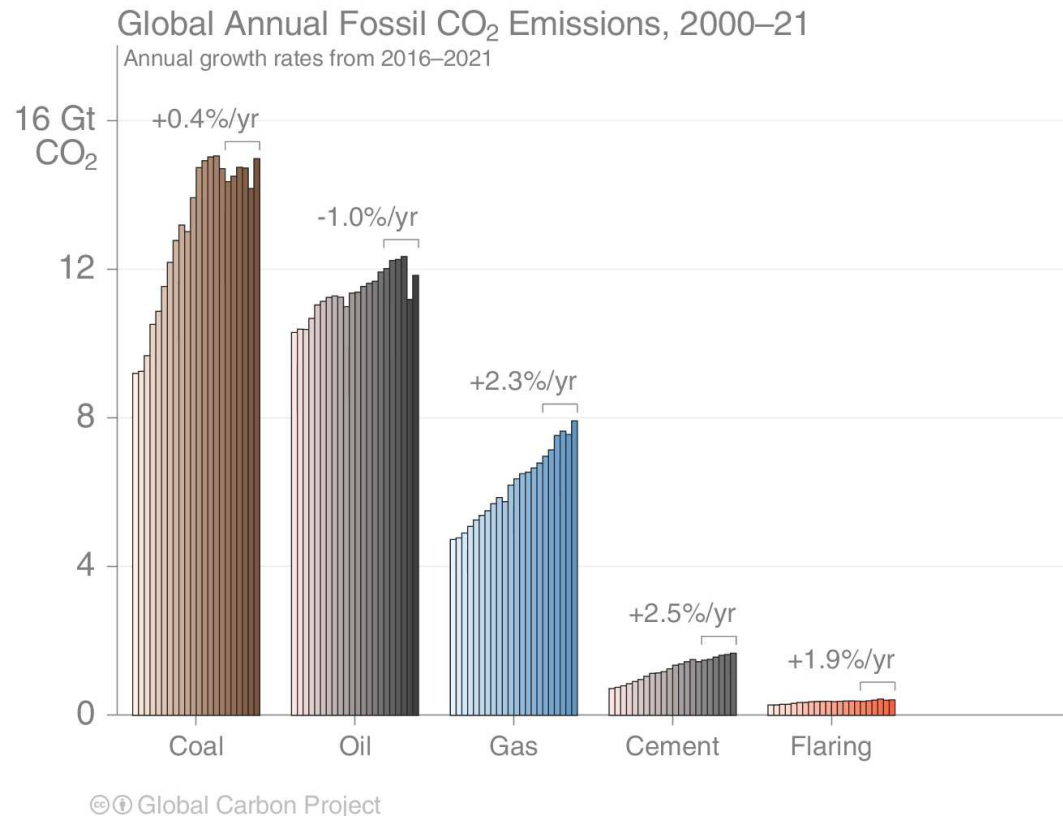
Emissioni globali di CO<sub>2</sub> fossile: 37.1 ± 2 GtCO<sub>2</sub> in 2021, 63% in più del 1990.  
 Proiezioni per il 2022: 37.5 ± 2 GtCO<sub>2</sub>, 1.0% [0.1% to +1.9%] più alto del 2021.



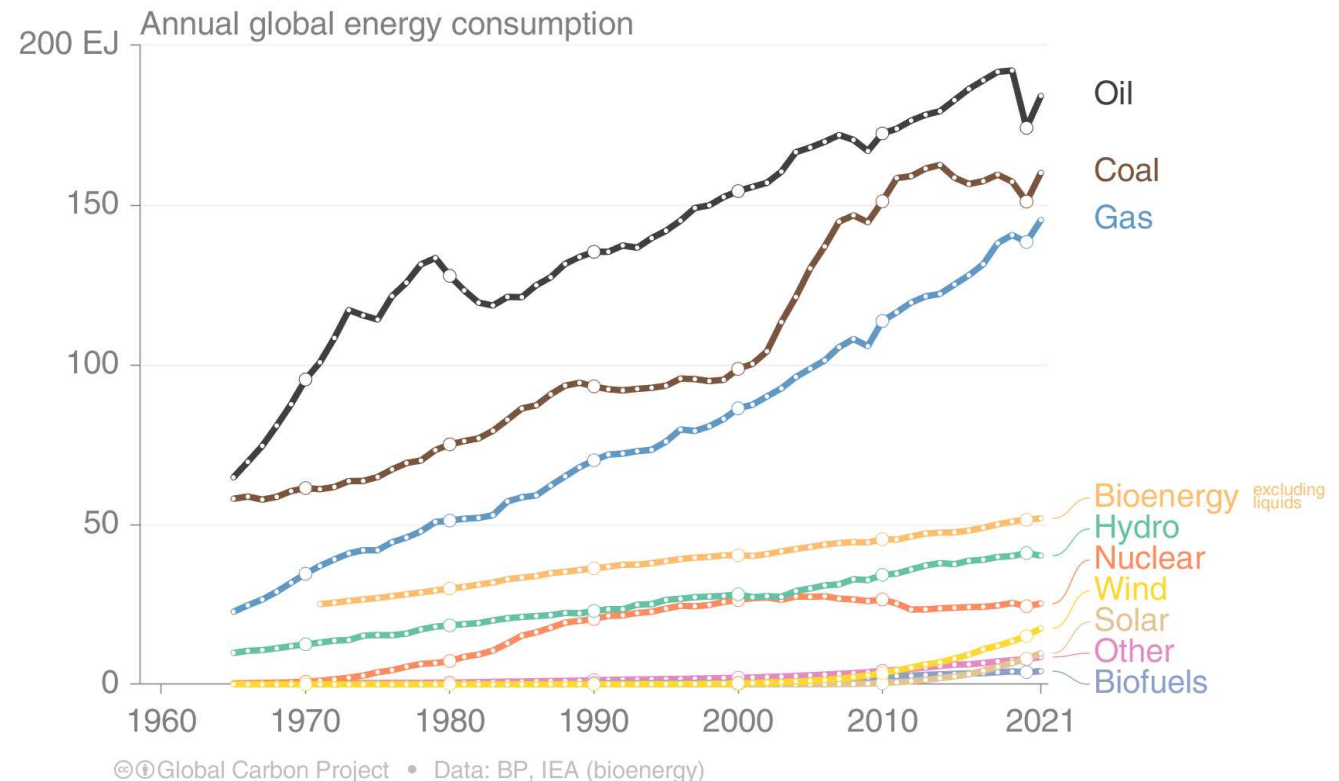
Source: Friedlingstein et al 2022; Global Carbon Project 2022

# FONTI DI EMISSIONE DI CO<sub>2</sub>

L'uso di carbone e petrolio è sceso complessivamente, tuttavia la crescita delle energia rinnovabili deve aumentare più velocemente per soddisfare il crescente fabbisogno e allo stesso tempo sostituire le fonti fossili.



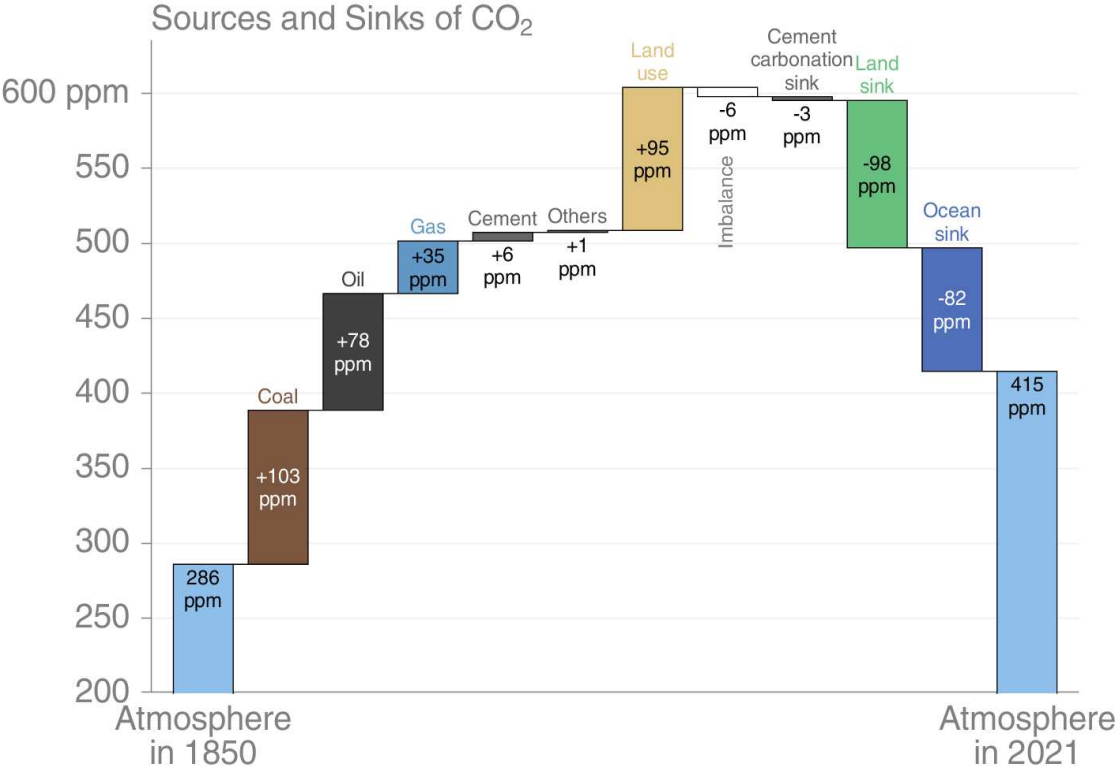
Source: CDIAC; Global Carbon Project 2022



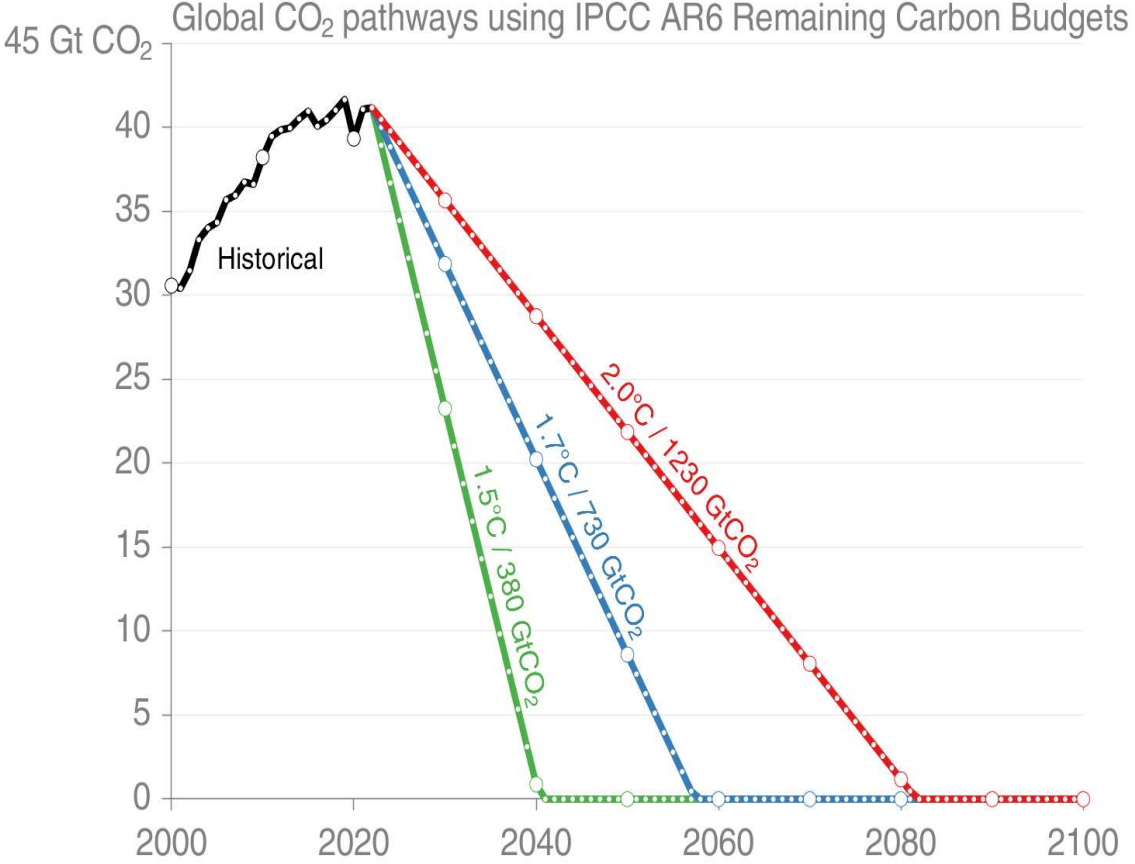
Source: BP 2022; Global Carbon Project 2022

# GLOBAL CARBON BUDGET

Il "carbon imbalance" rappresenta un divario nella attuale comprensione tra fonti e serbatoi di CO<sub>2</sub>.



© Global Carbon Project



© Global Carbon Project

Source: Friedlingstein et al 2022; Global Carbon Project 2022



# CARBON CAPTURE UTILISATION AND STORAGE

## Capture

Capturing CO<sub>2</sub> from e.g. biomass-fuelled power stations, industrial facilities or directly from air.



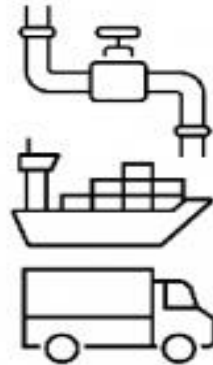
## Use

Captured CO<sub>2</sub> is used as a resource or feedstock to create products and services.



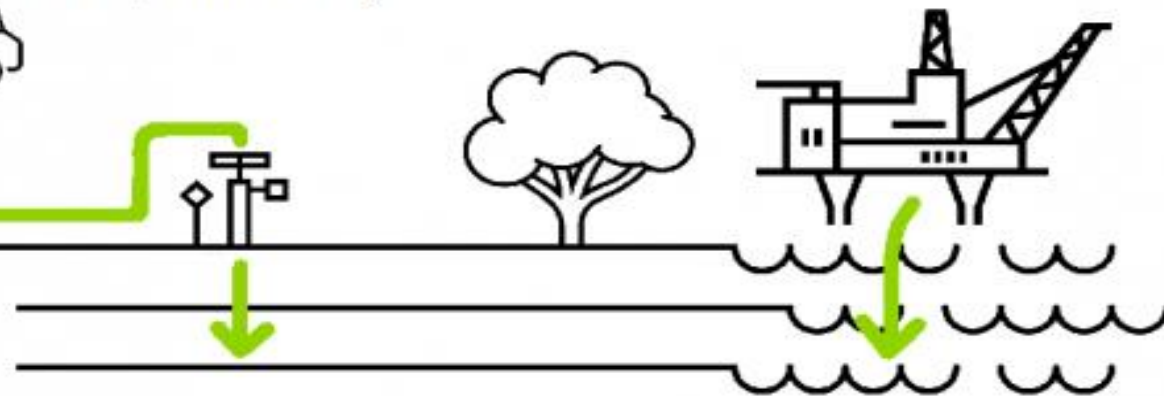
## Transport

The compressed CO<sub>2</sub> is moved by ship, truck or pipeline from point of capture to the point of use or storage.



## Storage

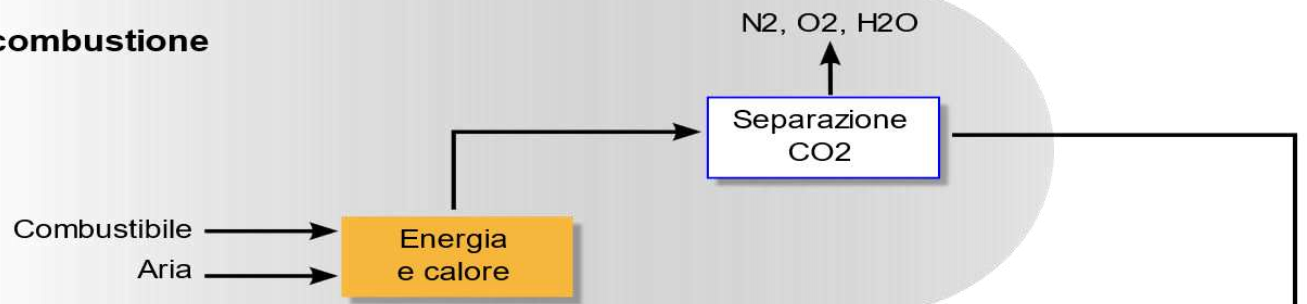
Storing the CO<sub>2</sub> permanently in underground geological formations.



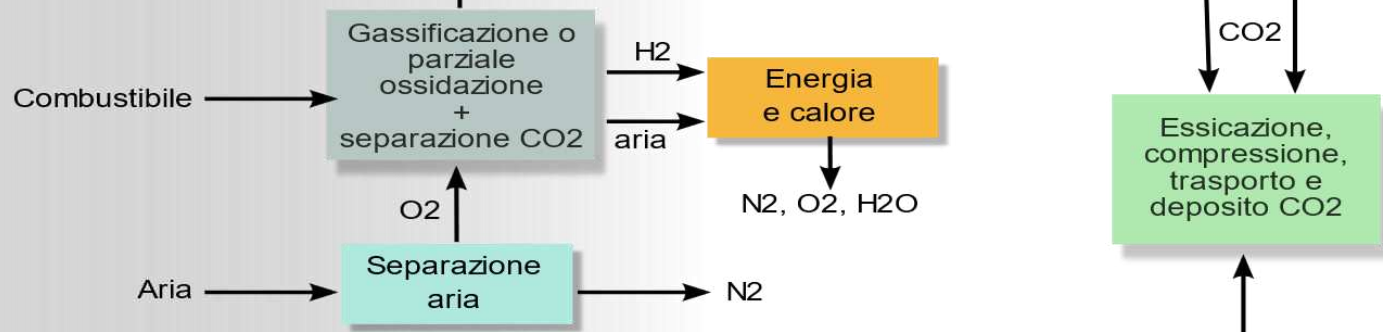
Source: <https://climate.ec.europa.eu>

# DIFFERENTI TIPI DI CATTURA

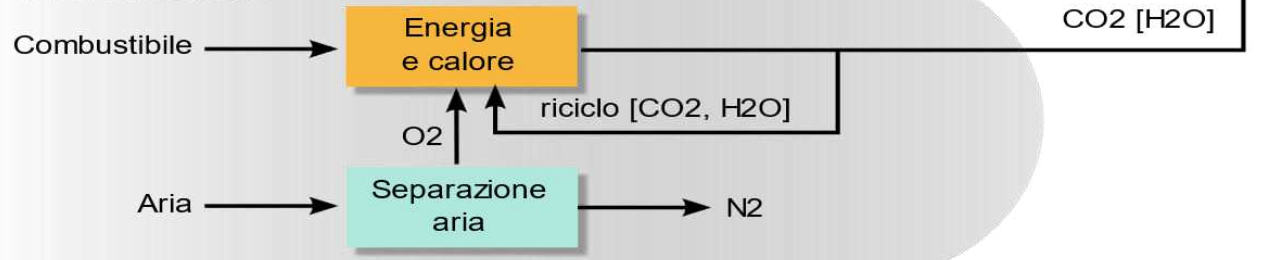
## Cattura post-combustione



## Cattura pre-combustione

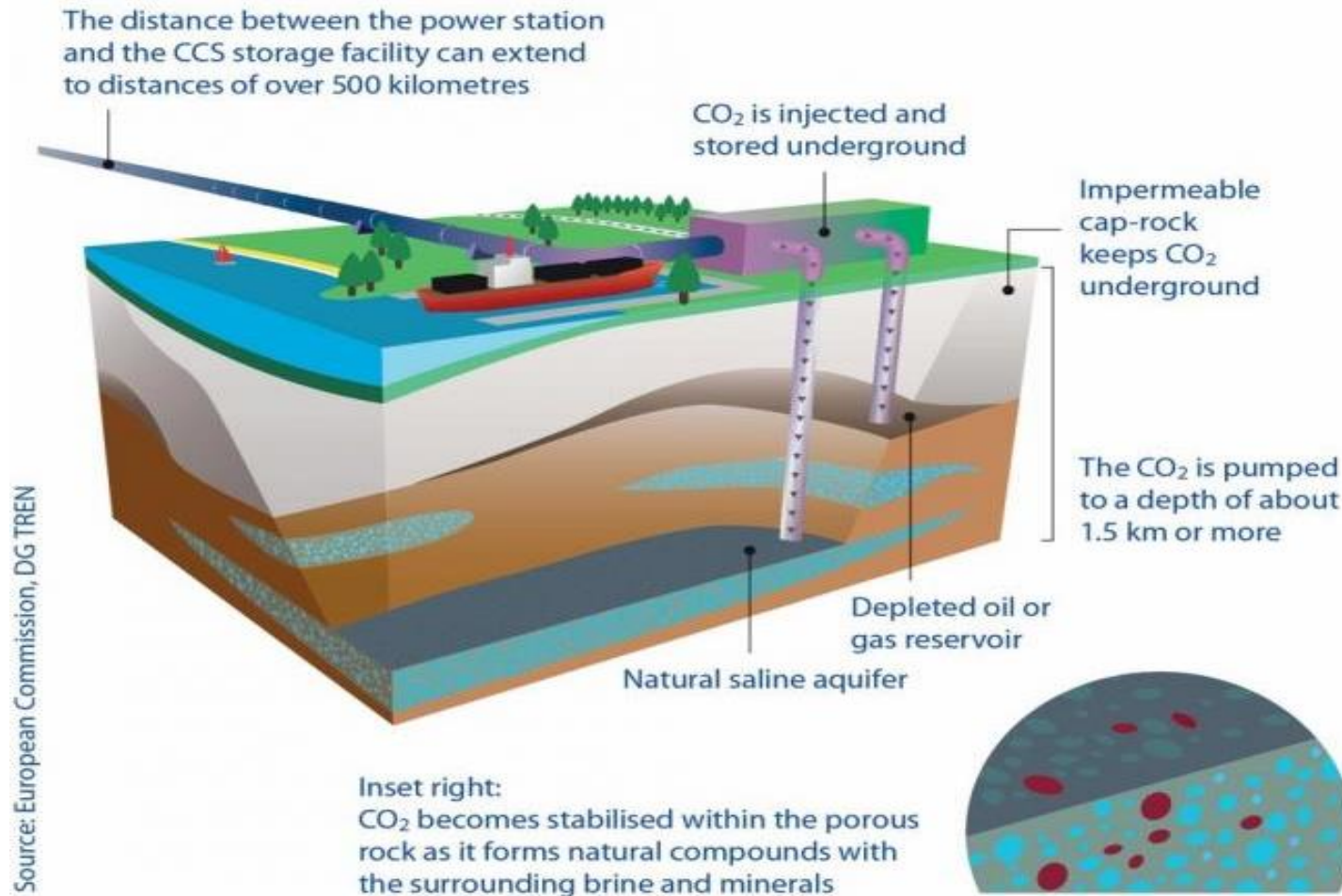


## Cattura per ossi-combustione



Sources: IEA, 2008b based on IPCC, 2005.

# CARBON CAPTURE AND STORAGE (CCS)



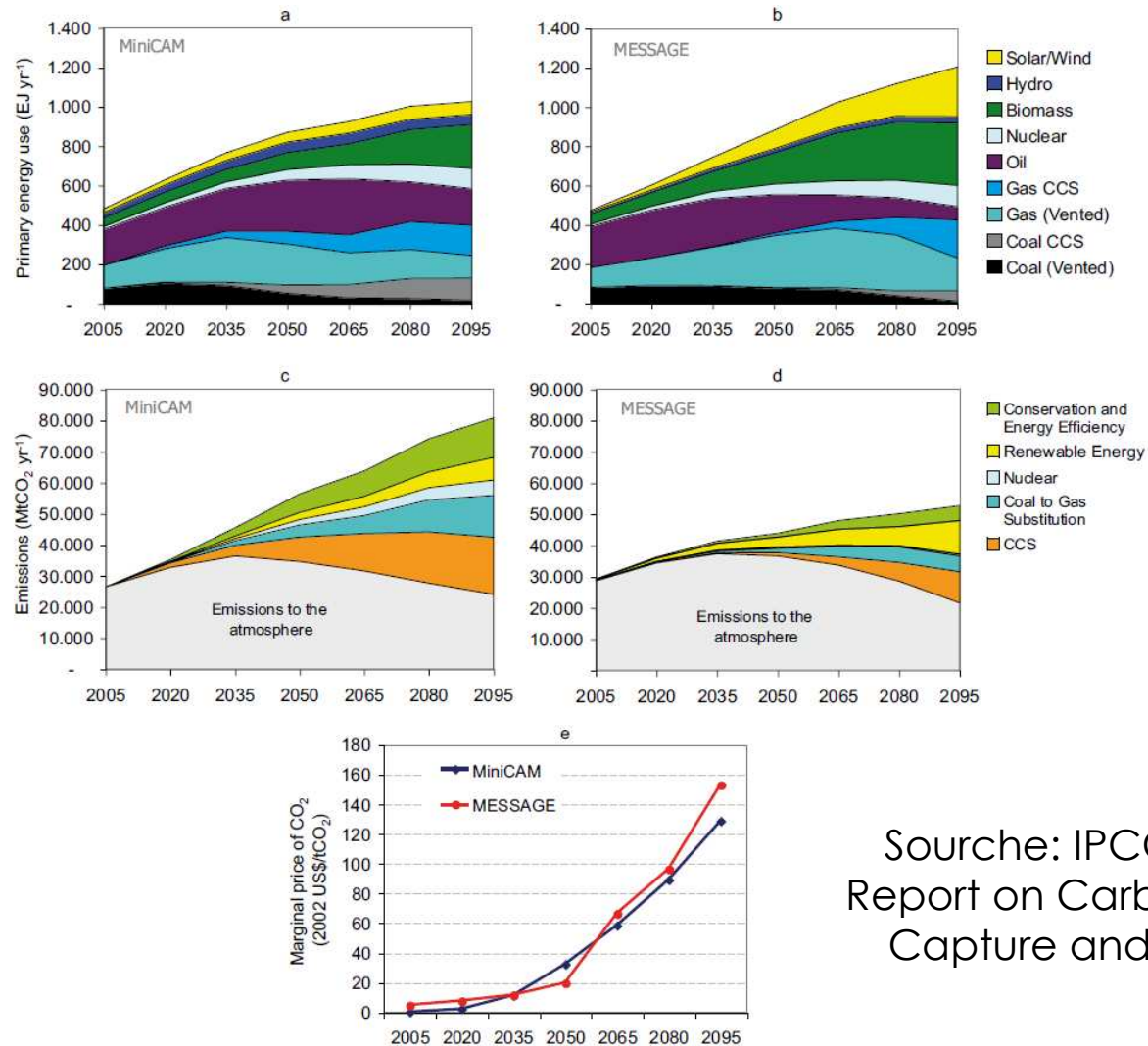
Source: European Commission, DG TREN

Principali obiettivi:

- assicurarsi che la CO<sub>2</sub> catturata e stoccata rimanga isolata dall'atmosfera nel lungo periodo;
- assicurarsi che cattura, trasporto e stoccaggio non presentino rischi per la salute umana e degli ecosistemi.



# CONTRIBUTO DEL CCS



Queste immagini sono un esempio illustrato del potenziale globale del CCS come parte di un più grande portfolio di misure di mitigazione.

Si basano su due modelli alternativi di valutazione integrate (MESSAGE e MiniCAM) che adottano le stesse assunzioni per i principali drivers di emissioni.

Sourche: IPCC Special Report on Carbon Dioxide Capture and Storage.

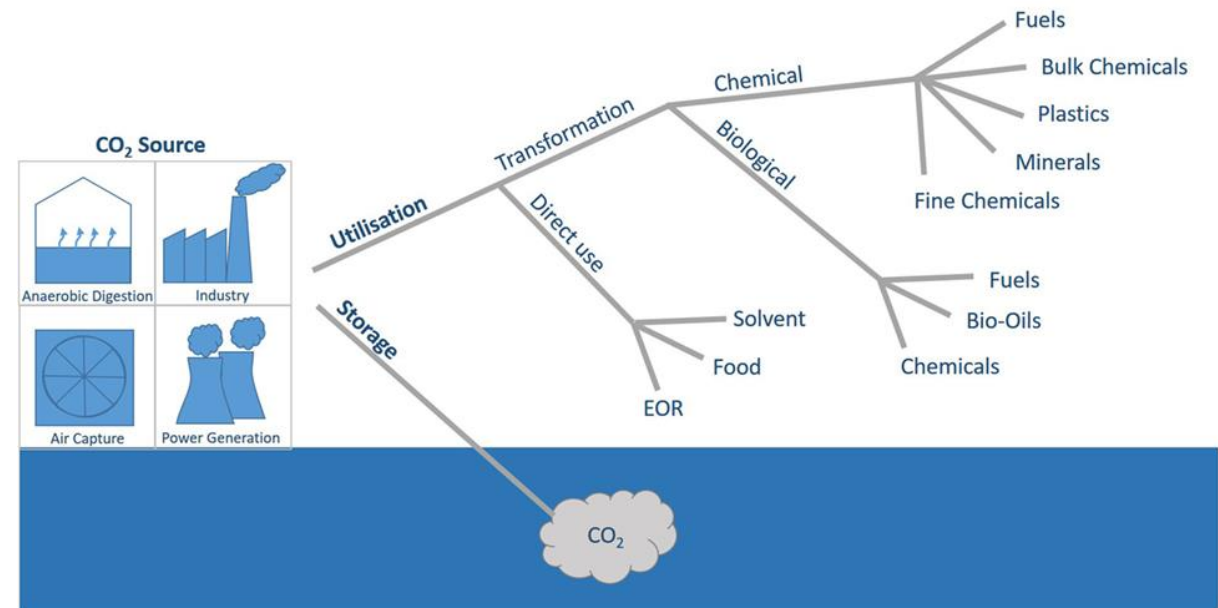


# CARBON CAPTURE AND UTILISATION (CCU)

L'utilizzo della CO<sub>2</sub> nei processi di produzione dipende dalle tecnologie e procedure usate:

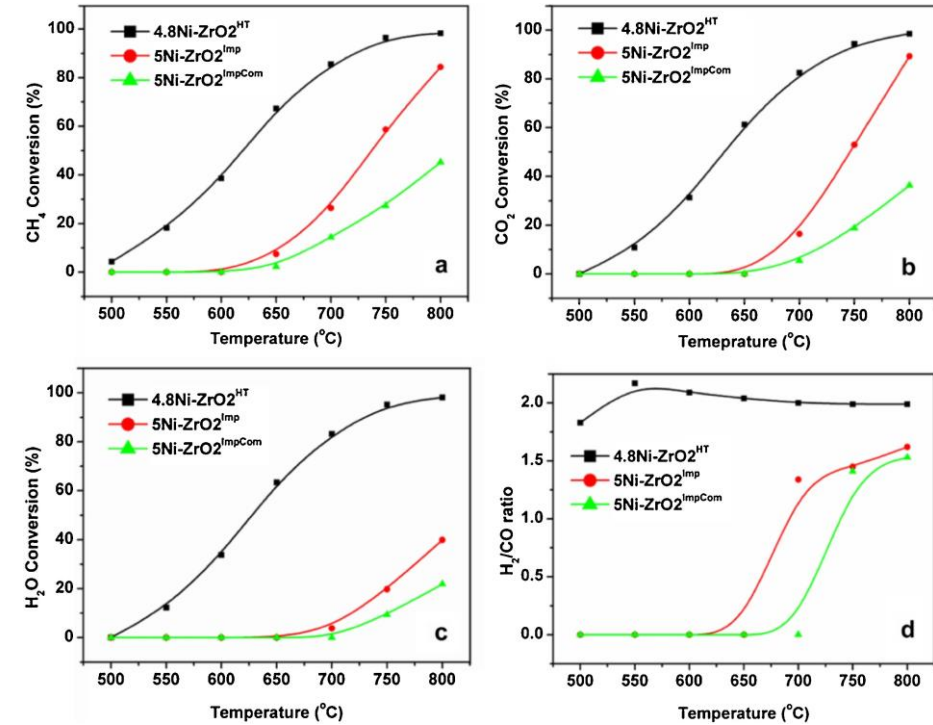
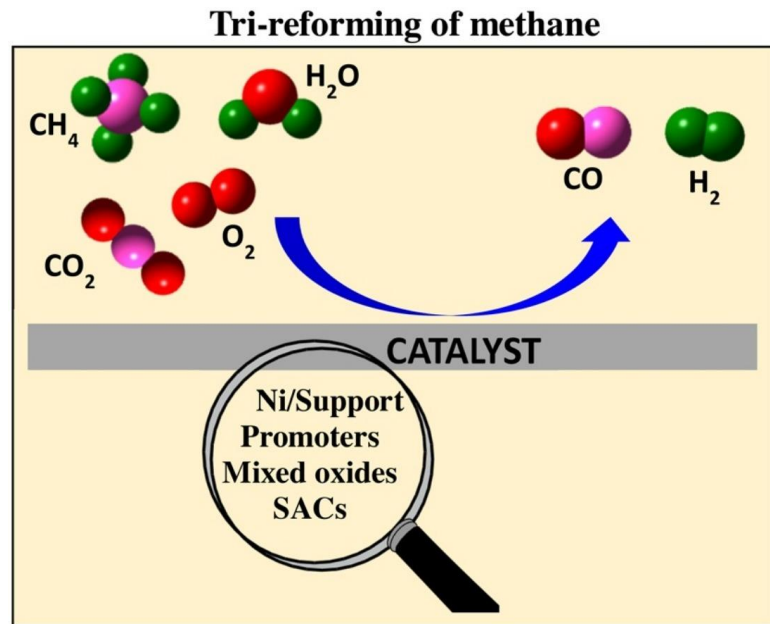
- uso diretto di CO<sub>2</sub> in bevande e serre;
- utilizzo come fluido di lavoro o solvente per Enhanced Oil Recovery (EOR);
- uso come materia prima e conversione in prodotti a valore aggiunto come polimeri, materiali da costruzione, prodotti chimici e combustibili sintetici.

Quest'ultima branca di nuove tecnologie che utilizzano la CO<sub>2</sub> come materia prima può contribuire all'economia circolare e agli obiettivi di mitigazione del clima.





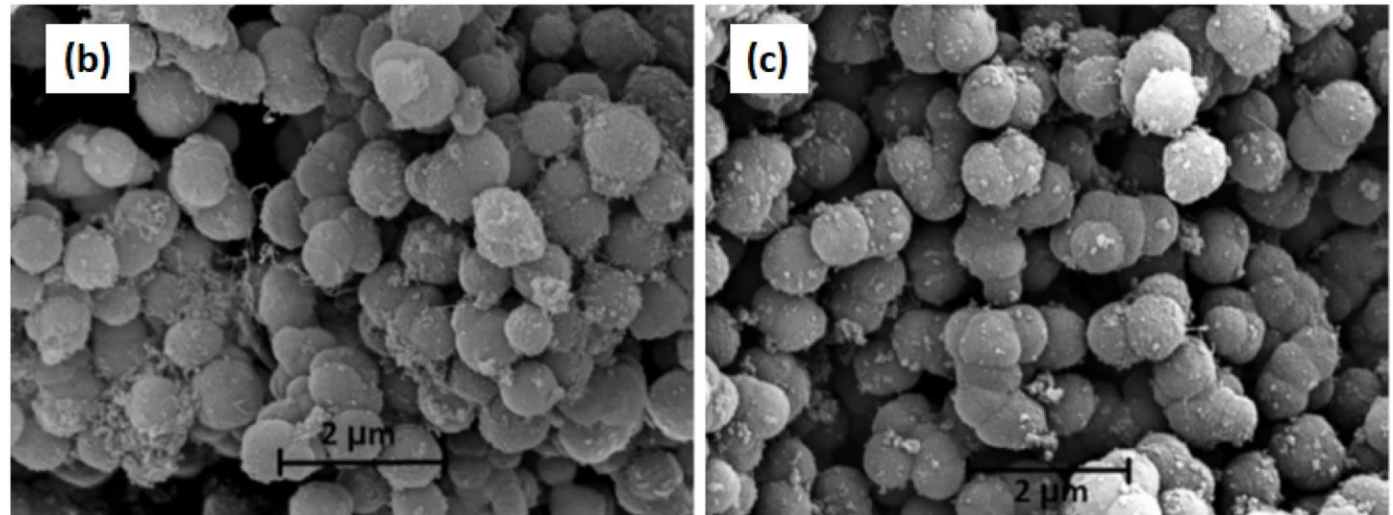
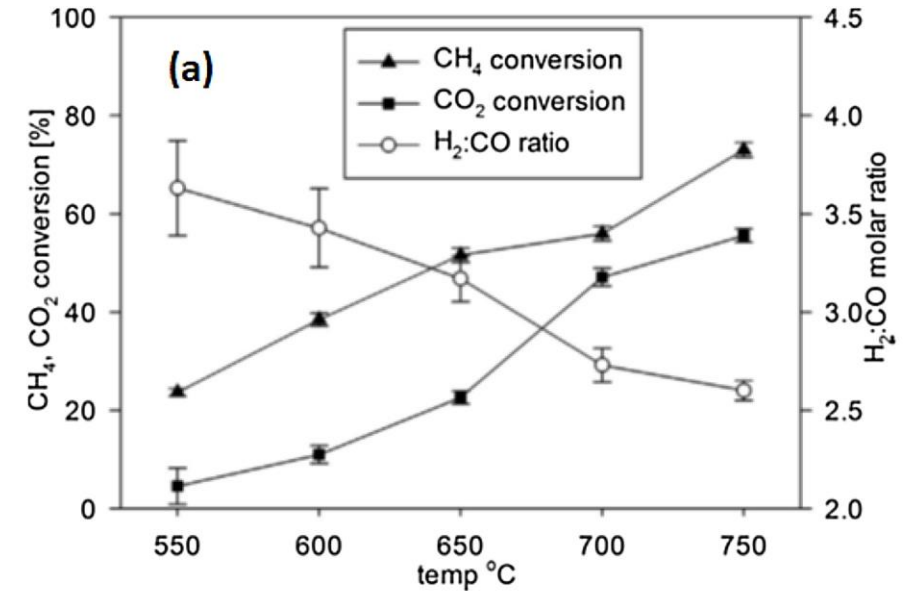
# TRI-REFORMING - CATALIZZATORI



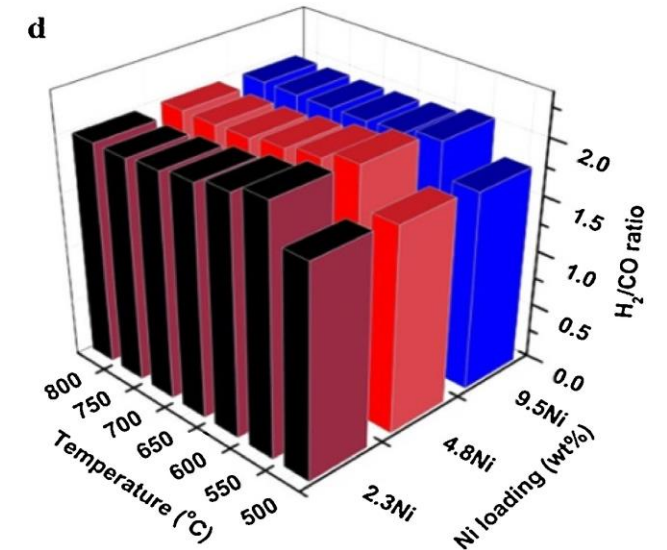
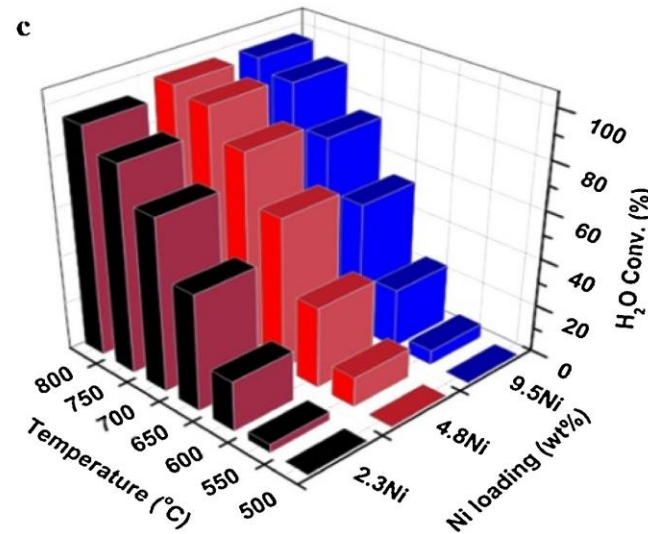
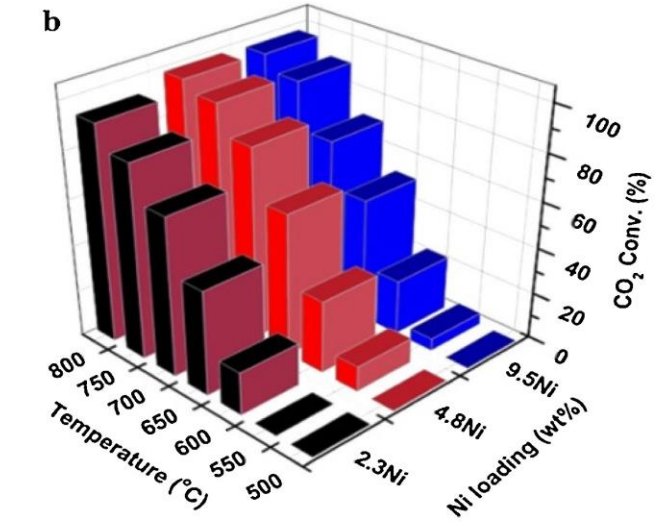
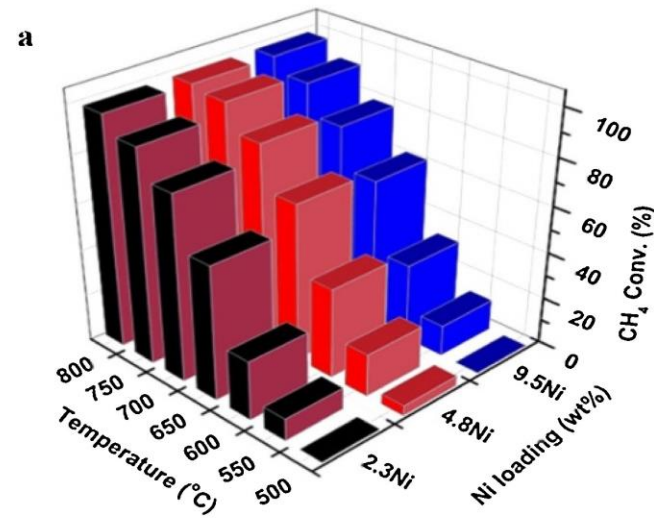
Effect of catalyst preparation method on (a) CH<sub>4</sub> conversion, (b) CO<sub>2</sub> conversion, (c) H<sub>2</sub>O conversions and (d) H<sub>2</sub>/CO ratios. Reaction conditions: reaction time = 6 h, temperature = 500–800 °C, GHSV = 80,000 mL g<sup>-1</sup> h<sup>-1</sup>, molar ratio of O<sub>2</sub>:CO<sub>2</sub>:H<sub>2</sub>O:CH<sub>4</sub>:He = 1:1:2.1:5:18.



Conversion of  $\text{CH}_4$  and  $\text{CO}_2$  (average value from 4 h reaction) over  $\text{Ni}/\text{SiO}_2$  catalyst at different temperatures. Reaction conditions: molar ratio of the feed  $\text{CH}_4:\text{CO}_2:\text{H}_2\text{O}:\text{O}_2:\text{He} = 1:0.5:0.5:0.1:0.4$ ,  $\text{CH}_4$  gas flow rate  $25 \text{ mL min}^{-1}$ , 0.2 g of catalyst. b) and c) Scanning Electron Microscopy (SEM) micrographs of the  $\text{Ni}/\text{SiO}_2$  core/shell catalyst after 4 h reaction at b)  $550 \text{ }^\circ\text{C}$ , and c)  $750 \text{ }^\circ\text{C}$ .



Effect of Ni loading during TRM over Ni-ZrO<sub>2</sub> on  
 (a) CH<sub>4</sub> conversion, (b) CO<sub>2</sub> conversion, (c) H<sub>2</sub>O conversion, and (d) H<sub>2</sub>/CO ratio.  
 Reaction conditions: reaction time = 6 h, temperature = 500–800 °C, GHSV = 80,000 mL g<sup>-1</sup> h<sup>-1</sup>, molar ratio of O<sub>2</sub>:CO<sub>2</sub>:H<sub>2</sub>O:CH<sub>4</sub>:He = 1:1:2.1:5:18.



Relative comparison of TRM catalyst supports on the basis of the nominated criteria, taking into consideration the operation conditions of this process.

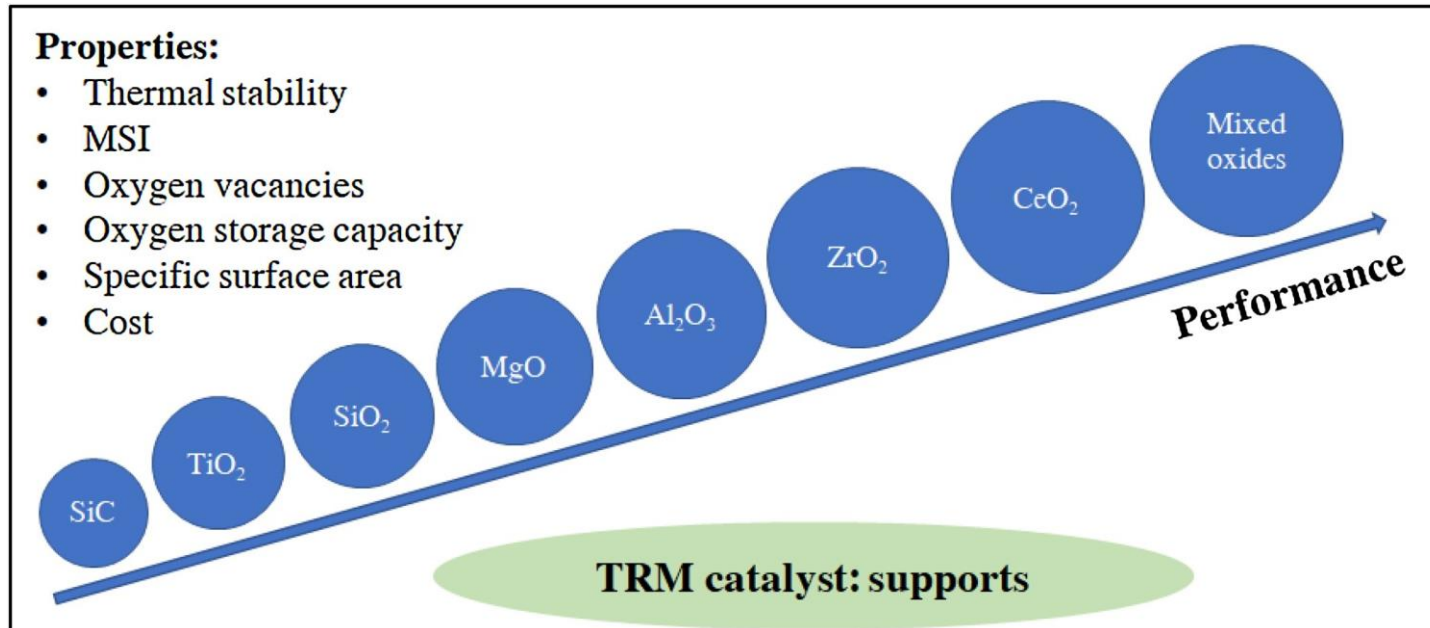
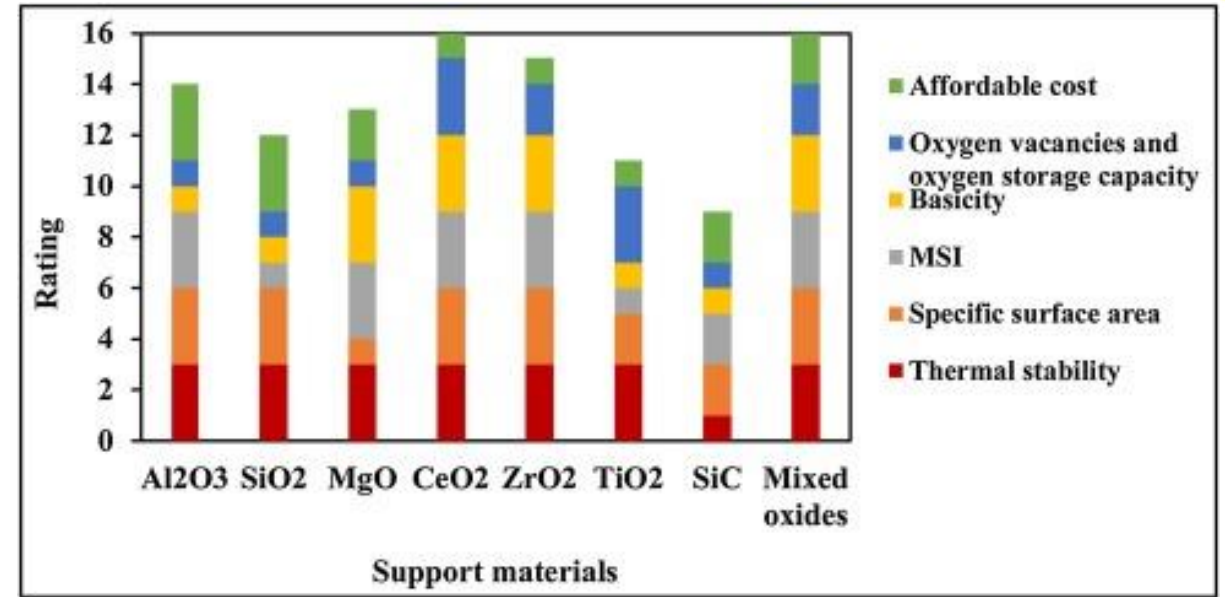
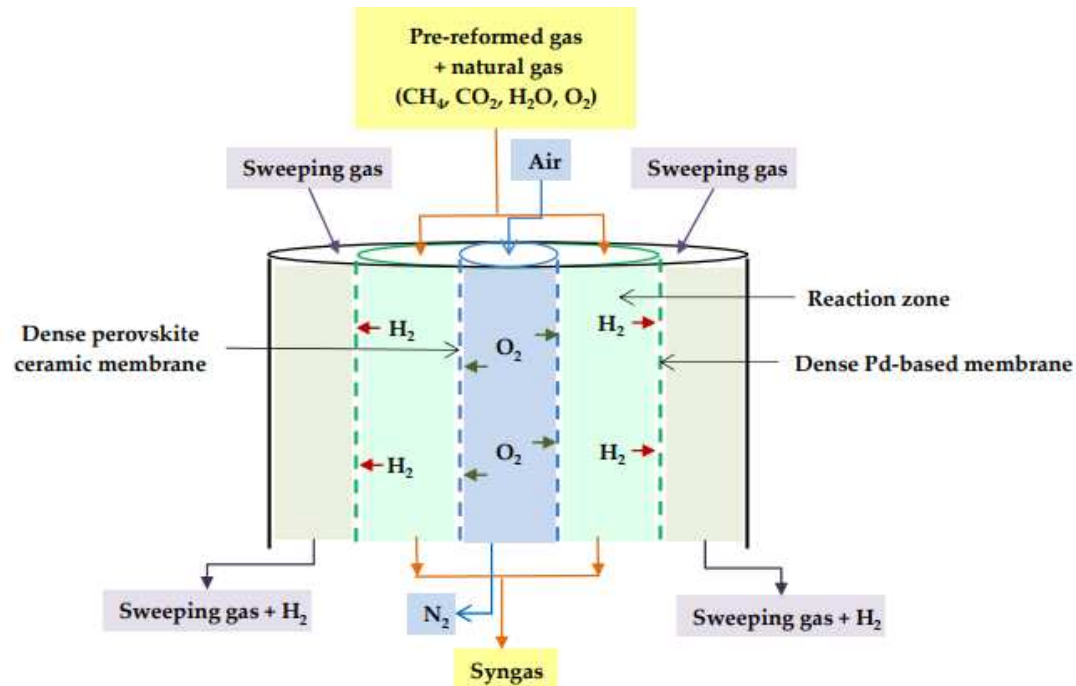


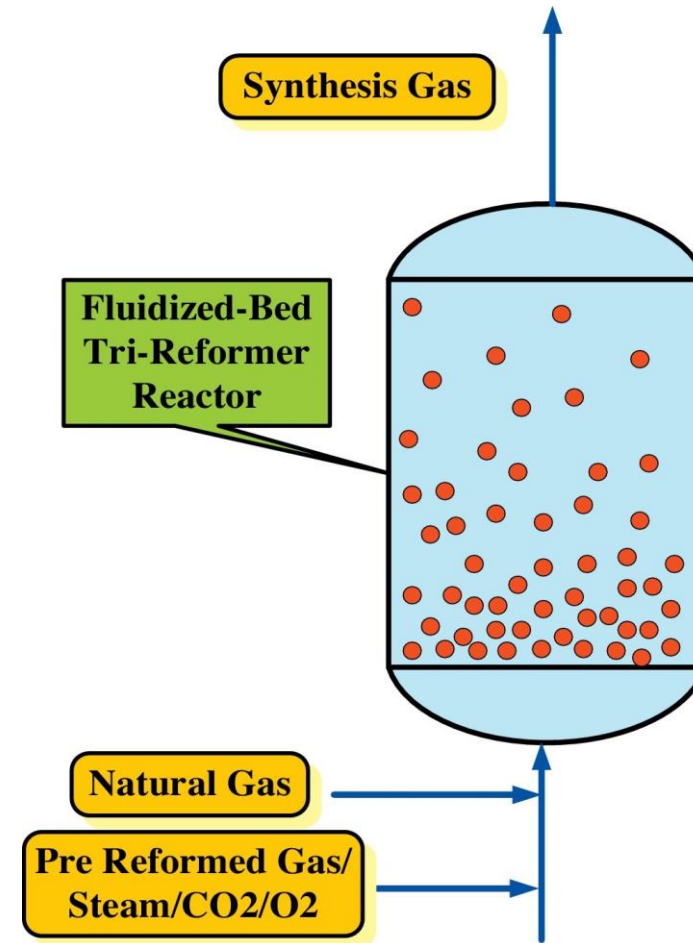
Illustration of the impact of supports in the design of TRM catalysts.



# TRI-REFORMING - REATTORI

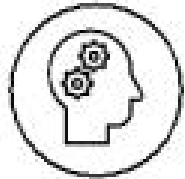


Permeable Membrane Reactor.



Schematic diagram of proposed novel fluidized-bed tri-reformer reactor.

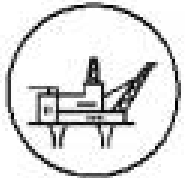
# OSTACOLI DA SUPERARE



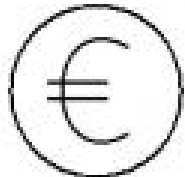
**Lack of technical expertise**  
due to long industry chain



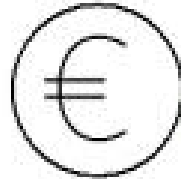
**Resource usage at scale**  
removing 1 Gtpa could consume as much as 1.4x the total electricity generation for the EU in 2018



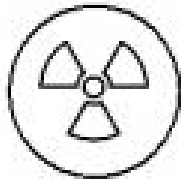
**Technology performance**  
CCS still in developing phase with uncertainty in performance



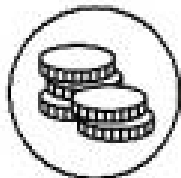
**CAPEX uncertainly**  
immaturity of CCUS & uniqueness of industrial sites limits replicability & cost estimation



**OPEX uncertainty**  
uncertainty of future oil prices



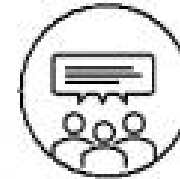
**Risk perception**  
investors require high ROI due to high risk



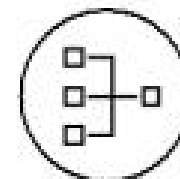
**Lack of revenue model**  
due to low CO<sub>2</sub> prices & insufficient utilisation opportunities



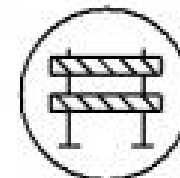
**Uncertainty in demand**  
for industrial products



**Public resistance**  
often because of lack of knowledge



**Policy uncertainty**  
lack of comprehensive frameworks and business models to facilitate CCUS



**Regulations & Infrastructure**  
admin causes 5-10 year delays and slow the pace of construction



**Cross-chain integration**  
many uncertainties in coordination between many stakeholders, volume risk of CO<sub>2</sub> & CO<sub>2</sub> reliability transfer

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