



Modelling of multistream heat exchanger for natural gas liquefaction

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Introduction

- "green" energy source, with market in expansion.
- Necessity to reduce transported volumes.
- LNG cost distribution.







Why modelling LNG HX

HX is the core equipment in the major 0.259
 LNG processes.

Intellectual property are highly guarded and not freely accessible.







Why modelling LNG HX

Traditional sizing tecnique are not applicable:

- Complex geometry
- Extreme operative conditions $T \rightarrow -160: 30 \ ^{\circ}C$

P → 1 : 70 bar

High efficiency of LNG heat exchanger (

$$\varepsilon = \frac{c_{p,in} \left(T_{C,out} - T_{C,in} \right)}{c_{p,\min} \left(T_{H,in} - T_{C,in} \right)} > 90\%)$$



Thesis objectives

- Development of a model for NG liquefaction HX:
 - Plate and Fin Heat Exchanger (PFHX);
 - Coil Wound Heat Exchanger (CWHX);



MAX: 1.5m·3.0m·8.2m



Diameter ~5m Length ~40m

Plate and Fin Heat Exchanger (PFHX)





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Model structure

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- Development of a general and flexible architecture able to simulate a wide range of operative conditions and designs.
- Development of a model usable by the costumer for the simulation of his LNG HX.
- The model should consider:
 - Specific geometry of the fins,
 - Changes in fluid properties,
 - Variable htc and Fanning factor.

Model structure





PFHX sub-models: offset fins

Shah and Sekulic, Fundamentals of Heat Exchanger Design, 2003



Geometry model

$$\begin{split} A_{frontal} &= L_1 \cdot L_2 \\ A_{cross,ch} &= h \cdot L_2 - (h-t) \cdot t \\ A_{surfacesch} &= 2 \cdot L_2 L_1 - 2 \cdot t \cdot L_1 + 2 \cdot h \cdot L_2 + 2(h+2t) \cdot L_2 + 2(h-t) \cdot L_1 \\ &+ 2(h-2t) \cdot t \cdot n_{offsetstrips} + (s+t)t \cdot (n_{offsetstrips} - 1) + 2(s+t)t \end{split}$$



HTC model

$$j(z) = 0.6522 \operatorname{Re}(z)^{-0.5403} \left(\frac{s}{h}\right)^{-0.1541} \left(\frac{t}{l_f}\right)^{0.1499} \left(\frac{t}{s}\right)^{-0.366} \cdot \left[1 + 5.269 \cdot 10^{-5} \operatorname{Re}(z)^{1.34} \left(\frac{s}{h}\right)^{0.504} \left(\frac{t}{l_f}\right)^{0.456} \left(\frac{t}{s}\right)^{-1.055}\right]^{0.1}$$

• Fanning factor model

$$f(z) = 9.6243 \operatorname{Re}(z)^{-0.7433} \left(\frac{s}{h-t}\right)^{-0.1856} \left(\frac{t}{l_f}\right)^{0.3053} \left(\frac{t}{s}\right)^{-0.2659} \left[1+7.66 \cdot 10^{-8} \operatorname{Re}(z)^{4.429} \left(\frac{s}{h-t}\right)^{0.92} \left(\frac{t}{l_f}\right)^{3.767} \left(\frac{t}{s}\right)^{0.236}\right]^{0.1}$$

Operative conditions.

	NG	MR vapour	MR liquid
Flowrate [kg/s]			
	118	200	400
Temperature [K]	\bigcirc		\frown
	305	305	(167)
Pressure [bar]			\sim
	66.5	38.6	9.8



Design

Tab: Design Hot streams				
Number of channels per stream	150 NG			
	700 MR			
Operation mode	Co-current			
Plate flow length	5	[m]		
Plate width	1.5	[m]		
Spacing between plates	0.665	[mm]		
Spacing between fins	5	[mm]		
Hydraulic radius	0.143	[mm]		
Fin thickness	0.0254	[mm]		

PFHX offset fins: results T, VF



PFHX offset fins: results HTC



Comparison HTC





Conclusions



The final model accounts for:

- fluid properties variation in the axial direction,
- specific geometry of the HX,
- fluid to wall variable convective heat transfer in the axial direction,
- variable friction factor in the axial direction,
- single phase correlations for htc(z) and f(z).

The model structure and the correlations used for the channel simulation have been usefully implemented in the PSE model libraries.



Evaluation of the model with experimental data.

Development of a 2-D model (especially for the CWHX).



Thanks for your attention!

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- Programme
- gPROMS Model builder 4.1.0.20140520 (x64) dev

Streams composition



Table A.1. NG stream composition used for the simulations.				
Nitrogen	0.0144	kg/kg		
Methane	0.8622	kg/kg		
Ethane	0.0534	kg/kg		
Propane	0.0328	kg/kg		
2-Methylpropane	0.0033	kg/kg		
Butane	0.0020	kg/kg		
2-Methylbutane	0.0033	kg/kg		
Pentane	0.0020	kg/kg		
Hexane	0.0049	kg/kg		
Heptane	0.0034	kg/kg		

Table A.2. MR stream composition used for the simulations.

Nitrogen	0.03822	kg/kg
Methane	0.4793	kg/kg
Ethane	0.4132	kg/kg
Propane	0.0024	kg/kg
2-Methylpropane	0.02728	kg/kg
Butane	0.03952	kg/kg